RSN

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Verified by A. Ristic, J. Grand

Type Material: K4Na12[Si28Zn8O72]·18H2O

Method: C. Röhrig and H. Gies [1]

Batch Composition: TEOS: 0.1 ZnO: 0.5 NaOH: 0.5 KOH: 0.08 TEAOH: 44 H₂O

Source Materials

deionized water (DI) zinc oxide (99%, Prolabo) sodium hydroxide (97%, Sigma Aldrich) potassium hydroxide (97%, Sigma Aldrich) tetraethylammonium hydroxide (35% water solution, Aldrich) tetraethoxysilane (98%, Sigma Aldrich)

Batch Preparation (for 2.9 g dry product)

(1) [53.434 g water + 2.078 g TEAOH + 1.44 g NaOH + 2.302 g KOH] stir in a polypropylene bottle until clear solution is formed
(2) [(1) + 0,574 g zinc oxide] stir for 30 min
(3) [(2) + 15 g tetraethoxysilane] hydrolyze for 8 h^a

Crystallization

Vessel: Teflon-lined stainless steel autoclave Temperature: 180° C Time: 10 days Agitation: none

Product Recovery

(1) Dilute reaction mixture with water^b

- (2) Filter and wash with water
- (3) Dry at 80°C

(4) Yield: 2.9 g

Product Characterization

XRD: RSN; competing phase: no Crystal size and habit: intergrown crystals forming large aggregates, size 100-1500 nm.

Reference

[1] C. Röhrig, H. Gies, Angew. Chem. Int. Ed. 34 (1995) 63 **Notes**

a. ZnO is dispersed in the silicate solution.

b. The product precipitates at the bottom of the autoclave. It is very hard to crush.